

Direct Optical Resolution of Vicinal Diols and an α -Hydroxy Oxime
with a Vicinal Diamine

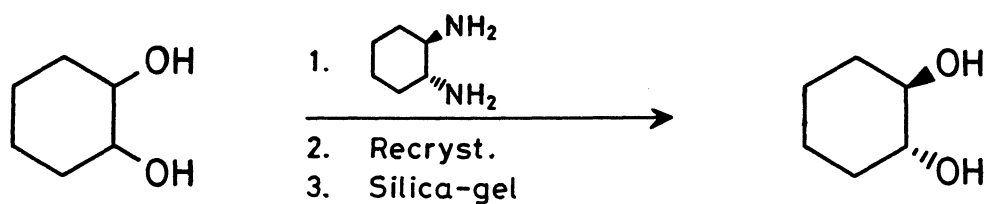
Masatoshi KAWASHIMA* and Akihisa HIRAYAMA

Kankyo Kagaku Center Co., Ltd., Kamariyacho, Kanazawa-ku, Yokohama 236

Vicinal diols, such as 1,2-cyclohexanediol, 1,2-cyclo-octanediol, 2,3-butanediol, and 1,2-diphenyl-1,2-ethanediol, were directly resolved with (1R,2R)-(-)-1,2-cyclohexanediamine (1). (E)-1,2-Diphenyl-2-(hydroxyimino)ethanol was also resolved efficiently with 1.

Although asymmetric syntheses of vicinal diols have been reported,¹⁾ direct optical resolution of vicinal diols has never been reported to our knowledge. In general, optical resolutions of vicinal diols have been carried out by multi-step procedures via the acylated derivatives.²⁾

Previously we reported³⁾ the direct optical resolution of 2,2'-dihydroxy-1,1'-binaphthyl with optically active 1,2-cyclohexanediamine (1), where the efficient resolution seemed to be due to the situation that the binaphthyl and 1 have the same structural symmetry of C_2 and further a pair of hydrogen-bonds are probably formed between them to favor the resolution. So we planned the resolution of compounds with vicinal alcoholic, instead of phenolic, dihydroxy groups.



When the mixture of (1R,2R)-(-)-1 ($[\alpha]_D^{24}$ -36.7° (c 4.14, H₂O)) 2.00 g (17.5 mmol) and racemic trans-1,2-cyclohexanediol (2a) 1.97 g (17.0 mmol) in benzene (4 cm³) was heated to give a homogeneous solution and allowed to cool to room temperature, crystalline precipitates were obtained. They were recrystallized twice from benzene (each 4 cm³). The ¹H-NMR and IR spectra of the crystalline compound showed that the compound consisted of (1R,2R)-(-)-1 and 2a (1:1).⁴⁾ Treatment of the solution of the crystalline compound in ethanol with silica-gel short column afforded (1R,2R)-(-)-2a ($[\alpha]_D^{23}$ -31.3° (c 0.02, H₂O)) 0.718 g (6.18 mmol) in a yield of 73% based on the enantiomer existing in the racemate. Optical purity of the resolved 2a was 67%.⁵⁾

The results of optical resolution of other vicinal diols, such as 1,2-cyclooctanediol (2b), 2,3-butanediol (2c), and 1,2-diphenyl-1,2-ethanediol (2d), with (1R,2R)-(-)-1 are summarized in Table 1.

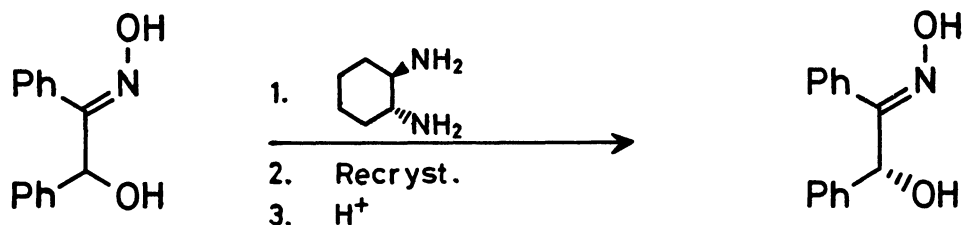
Table 1. Optical Resolution of Vicinal Diols with (1R,2R)-(-)-1^{a)}

| Diol | Resolved enantiomer | Chemical ^{b)} yield/% | $[\alpha]_D^{20}/^\circ$ | Optical ^{c)} purity/% |
|-----------|---------------------|--------------------------------|--|--------------------------------|
| <u>2b</u> | (1R,2R) | 36 | -14.8 (c 1.33, EtOH) | 88 |
| <u>2c</u> | (2R,3R) | 16 | -9.16 (c 1.09, CH ₃ COCH ₃) | 74 |
| <u>2d</u> | (1S,2S) | 62 | -90.4 (c 2.50, EtOH) | 91 |

a) All optical resolutions were carried out in the same manner as described in the text except for the isolation of optically active 2d: The crystalline compound, which consisted of (1R,2R)-(-)-1 and (1S,2S)-(-)-2d, was treated with dilute hydrochloric acid, and (1S,2S)-(-)-2d was extracted with CH₂Cl₂. b) Chemical yields were based on the enantiomer in the racemate. c) Optical purities were calculated from the optical rotations based on the reported values of $[\alpha]_D$ in the literature.⁶⁾

Moreover, optical resolution of an α -hydroxy oxime, which also has two hydroxy groups, with optically active diamine is of interest. Though α -hydroxy oximes are useful compounds in organic syntheses,⁷⁾ their optical resolution has not yet been reported.

Optical resolution of α -hydroxy oxime (3) was performed in the same manner as in the cases of diols described above. When the mixture of (1*R*,2*R*)-(-)-1 2.00 g (17.5 mmol) and racemic (*E*)-1,2-diphenyl-2-(hydroxyimino)ethanol (3) 4.00 g (17.6 mmol) in benzene (10 cm³) was heated to homogeneous solution and allowed to cool to room temperature, crystalline precipitates were obtained, which were twice recrystallized from benzene (each 50 cm³). The ¹H-NMR and IR spectra of the purified crystalline compound showed that the compound was composed of (1*R*,2*R*)-(-)-1 and 3 (1:1).⁸⁾ Then the crystalline compound was treated with 1 mol dm⁻³ hydrochloric acid and ethanol to afford (*R*)-(-)-(*E*)-3 ($[\alpha]_D^{23}$ -5.12° (c 0.508, CHCl₃)) 1.11 g (4.90 mmol) in a yield of 56% based on the enantiomer in the racemate. Optical rotation of the resolved 3 was larger than the reported value.⁹⁾



As described above, C₂-symmetrical vicinal diols (2) and α -hydroxy oxime (3) were directly and conveniently resolved with C₂-symmetrical vicinal diamine (1). The novel optical resolutions offer important methods for the acquisitions of optically active vicinal diols and α -hydroxy oximes.

References

- 1) For example, E. J. Corey, P. D. Jardine, S. Virgil, P. -W. Yuen, and R. D. Connell, J. Am. Chem. Soc., 111, 9243 (1989); E. N. Jacobsen, I. Marko, W. S. Mungall, G. Schroder, and K. B. Sharpless, *ibid.*, 110, 1968 (1988); M. Tokles and J. K. Snyder, Tetrahedron Lett., 27, 3951 (1986);

- M. Imuta and H. Ziffer, *J. Org. Chem.*, **43**, 3319 (1978); T. Oishi and M. Hirama, *ibid.*, **54**, 5834 (1989).
- 2) H. Brunner, U. Obermann, and P. Wimmer, *Organometallics*, **8**, 821 (1989); K. Laumen, D. Breitgoff, R. Seemayer, and M. P. Schneider, *J. Chem. Soc., Chem. Commun.*, **1989**, 148; J. P. Praly and G. Descotes, *Tetrahedron Lett.*, **23**, 849 (1982); M. V. Ganey, R. E. Padykula, G. A. Berchtold, and A. G. Braun, *J. Org. Chem.*, **54**, 2787 (1989); T. B. Grindley and R. Thangarasa, *J. Am. Chem. Soc.*, **112**, 1364 (1990); A. W. Ingersoll, *Org. React.*, **2**, 376 (1944); N. A. B. Wilson and J. Read, *J. Chem. Soc.*, **1935**, 1269.
- 3) M. Kawashima and A. Hirayama, *Chem. Lett.*, **1990**, 2299.
- 4) (1R,2R)-(-)-**1**•**2a** (1:1): mp 61-64 °C; $^1\text{H-NMR}(\text{CDCl}_3)$ δ 0.7-2.7 (m, 18H), 2.5 (brs, 6H), 3.1-3.6 (m, 2H); IR(KBr) 3450, 3400, 3350, 3600-2400, 1070 cm^{-1} ; $[\alpha]_{\text{D}}^{20}$ -28.0° (c 0.1, EtOH).
- 5) (1R,2R)-(-)-**2a**, $[\alpha]_{\text{D}}$ -46.5° (c 0.02, H₂O), D. M. Jerina, H. Ziffer, and J. W. Daly, *J. Am. Chem. Soc.*, **92**, 1056 (1970).
- 6) (1S,2S)-(+)-**2b**, $[\alpha]_{\text{D}}^{22}$ +16.9° (c 1.33, EtOH), E. J. Corey and J. I. Shulman, *Tetrahedron Lett.*, **1968**, 3655; (2S,3S)-(+)-**2c**, $[\alpha]_{\text{D}}^{31}$ +12.3° (c 3.5, CH₃COCH₃), K. A. Thaker and N. S. Dave, *J. Sci. Ind. Res., Sec. B*, **20**, 329 (1961); (1S,2S)-(-)-**2d**, $[\alpha]_{\text{D}}^{22}$ -99.1° (c 1.1, EtOH), J. Brugidou, H. Christol and R. Sales, *Bull. Soc. Chim. Fr.*, **1974**, 2033.
- 7) R. Bartnik, Y. Diab, and A. Laurent, *Tetrahedron*, **33**, 1279 (1977); F. A. Davis, M. S. Haque, and R. M. Przeslawski, *J. Org. Chem.*, **54**, 2021 (1989); T. Kudo, S. Kaizaki, J. Hidaka, and Y. Shimura, *Bull. Chem. Soc. Jpn.*, **48**, 2969 (1975).
- 8) (1R,2R)-(-)-**1**•**3** (1:1): mp 85-100 °C; $^1\text{H-NMR}(\text{CDCl}_3\text{-CD}_3\text{SOCD}_3)$ δ 0.7-2.7 (m, 10H), 2.7-4.6 (brs, 6H), 5.6 (s, 1H), 7.2 (s, 10H); IR(KBr) 3500, 3450, 3360, 3330, 3300, 3250, 3600-2400, 960, 710 cm^{-1} ; $[\alpha]_{\text{D}}^{20}$ -13.7° (c 1.03, CHCl₃).
- 9) (R)-(-)-(E)-**3**, $[\alpha]_{\text{D}}$ -3.2° (c 0.5, CHCl₃), K. Harada and S. Shiono, *Bull. Chem. Soc. Jpn.*, **57**, 1040 (1984).

(Received February 2, 1991)